

(relative intensity) 264 (5), 134 (48), 130 (14), 81 (100). Anal. Calcd for C_9H_8STe : C, 36.70; H, 2.31; S, 12.25; Te, 48.74. Found: C, 36.47; H, 2.43; S, 12.42; Te, 48.60.

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Extension of the Nenitzescu Reaction to a Cyclic Enamino Ketone. One-Step Synthesis of 6-Hydroxy-9H-pyrimido[4,5-b]indole-2,4-dione

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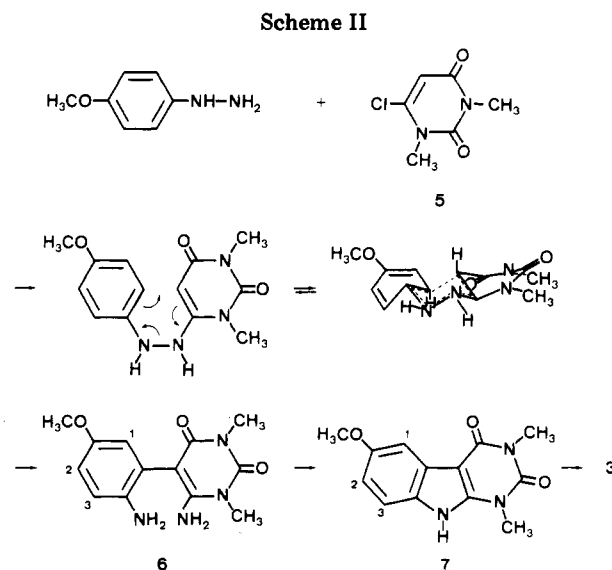
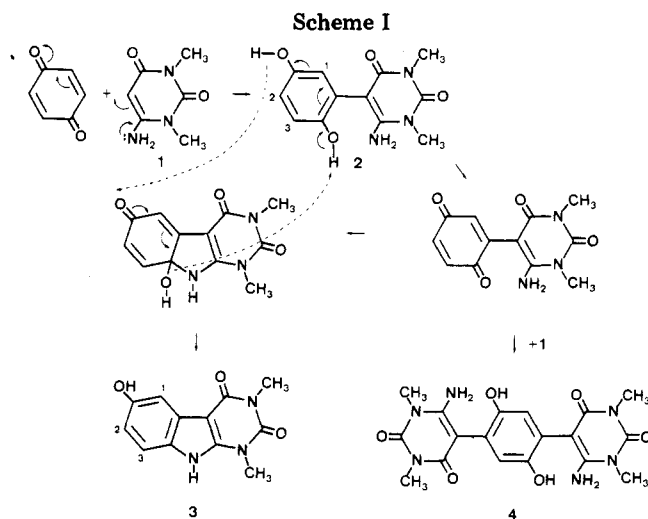
A new one-step synthesis of the 6-hydroxypyrimido[4,5-b]indole ring involving the condensation of *p*-benzoquinone with 1,3-dimethyl-6-aminouracil has been reported. This procedure, an extension of the Nenitzescu reaction, was compared to a more classical Fischer-type cyclization. In both reactions, intermediates have provided useful mechanistic information. When nitromethane was used as a solvent, this new Nenitzescu-type synthesis conclusively appeared to present some advantages, including simplicity and a relatively good yield.

The study of polycyclic DNA-intercalative drugs has been a matter of significant interest for several years. Recently, we have described an approach to a number of 6-hydroxycarbazoles¹ related to the well-known antileukemic 9-hydroxyellipticine.² We have also investigated the biological properties of 5-acyl-6-aminouracil derivatives in relation to their structural features.³

These initial findings prompted us to include the pyrimidine moiety in a tricyclic system in order to enhance both intercalative and antitumor properties. For this purpose, we have carried out a one-step synthesis of 6-hydroxy-9H-pyrimido[4,5-b]indole-2,4-dione starting from 1,3-dimethyl-6-aminouracil (1) and *p*-benzoquinone (PBQ) and using either acetic acid or nitromethane as the solvent. In fact, three compounds have been isolated: the hydroquinone 2, the expected product 3, and the diadduct 4 (see Scheme I).

This reaction could be presented as a novel application to the cyclic enamine 1 of the Nenitzescu reaction⁴ whose extension to aromatic amines has been recently described.^{1a} As demonstrated for classical hydroxyindole synthesis, the reaction presumably proceeds via a Michael addition followed by several steps including an internal oxidation-reduction.⁵

The initial carbon-carbon attack has been clearly demonstrated by the formation of a stable intermediate 2 and



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of the diadduct 4, which is in accordance with previously proposed mechanisms.⁶ Furthermore, it is noteworthy

that better yields of carbazole (41%) have been obtained by refluxing in nitromethane as described by Patrick and Saunders.^{5j}

The structures of the products were assigned on the basis of their spectral properties, and the nature of pyrimido[4,5-*b*]indole 3 was further confirmed by comparison with an independently synthesized sample. For this purpose, the pyrimido[4,5-*b*]indole ring could be obtained either by photolysis or thermolysis of 5-diaza-6-anilino-uracils^{1c,7} or by thermal catalyzed conversion of 6-phenylhydrazinouracils.⁸ This latter Fischer-type cyclization was chosen. Thus, the reaction of 1,3-dimethyl-6-chlorouracil (5) with *p*-methoxyphenylhydrazine in the presence of sodium carbonate led to the diamine 6. The obtention of such a C-C adduct during the course of the Fischer indole synthesis was initially presented as a nucleophilic attack on the C-5 site of the uracil ring by a ortho phenyl carbon, after N-protonation.⁹ The formation of 6 most likely proceeds via a [3,3] sigmatropic shift (Scheme II). Such a concerted mechanism seems favored by the electron-donating effect of the methoxy group.

Cyclization of 6 to 7 readily occurred by refluxing either in 98% formic acid or in dimethylaniline. Heating the 9*H*-pyrimido[4,5-*b*]indole 7 at 200 °C with pyridinium chloride afforded 3 in an overall yield of 30%.

The facility with which this new one-step cyclization to 6-hydroxy-9*H*-pyrimido[4,5-*b*]indole-2,4-diones occurs makes this type of approach particularly attractive for the synthesis of some polycyclic rings including dioxypyrimidine and hydroxyindole moieties, with a view toward making new molecules with potential biological activity.

Experimental Section

Standard Techniques. Infrared spectra were recorded with a Perkin-Elmer 177 infrared spectrometer using a potassium bromide pellet. NMR spectra were recorded on a JEOL JNM-FX100 spectrometer with tetramethylsilane as an internal standard. Electron-impact mass spectra were obtained with a quadrupole mass spectrometer (Ribermag R10-10, combined with Riber 400 data system) at 70 eV by using direct insertion. Chemical-ionization mass spectrum was taken by using methane as the reagent gas. High-pressure liquid chromatography (HPLC) was carried out on a Waters 6000A solvent delivery system equipped with a U6K injector and a dual detector system (UV at 2537 Å and a refractive index detector). A reverse-phase Nucleosil 7 C₁₈ column was used. Combustion analyses were performed on a Perkin-Elmer CHN240 apparatus. All new compounds exhibited consistent spectral data and elemental analyses.

Neitzescu-Type Synthesis. Refluxing 15.5 g of 1,3-dimethyl-6-aminouracil (1) and 10.8 g of *p*-benzoquinone in acetic acid (route A) or in nitromethane (route B) for 12 h yielded 2 (A, 25%; B, traces), 4 (A, 15%; B, 17%), and 3 (A, 8%; B, 41%).

The compounds were separated through fractional crystallizations or better by preparative HPLC, with methanol as a solvent in both cases.

1,3-Dimethyl-5-(2,5-dihydroxyphenyl)-6-aminouracil (2): mp 270 °C; CI mass spectrum (CH₄), *m/e* (relative intensity); 292 (M⁺ + 29, 11.9), 264 (M⁺ + 1, 100), 263 (M⁺, 59), 245 (6.9); EI mass spectrum, *m/e* (relative intensity): 263 (52.1), 246 (55.5), 245 (18.8), 189 (47.8), 57 (100); 100-MHz ¹H NMR (Me₂SO-*d*₆) δ 3.20 (s, 3 H, NCH₃), 3.35 (s, 3 H, NCH₃), 5.90 (s, 2 H, NH₂), 8.00 and 8.55 (s, 1 H, OH), 6.45 (d, 1 H, H-1, *J*_{H₁-H₂} = 1 Hz), 6.55 (dd, 1 H, H-2, *J*_{H₁-H₂} = 1 Hz, *J*_{H₂-H₃} = 8 Hz), 6.65 (d, 1 H, H-3, *J*_{H₂-H₃} = 8 Hz); IR (KBr) 3440, 3350, 3250, 1700 cm⁻¹. Anal. Calcd for C₁₂H₁₃N₃O₄: C, 54.75; H, 4.97; N, 15.96. Found: C, 54.81; H, 5.04; N, 15.76.

1,3-Dimethyl-6-hydroxy-9*H*-pyrimido[4,5-*b*]indole-2,4-dione (3): mp >300 °C; EI mass spectrum, *m/e* (relative intensity) 245 (M⁺, 100), 188 (60), 173 (21.8), 160 (51.2); 100-MHz ¹H NMR (Me₂SO-*d*₆) δ 3.30 (s, 3 H, NCH₃), 3.50 (s, 3 H, NCH₃), 12.10 (s, 1 H, OH), 7.35 (d, 1 H, H-1, *J*_{H₁-H₂} = 1 Hz), 6.75 (dd, 1 H, H₂, *J*_{H₁-H₂} = 1 Hz, *J*_{H₂-H₃} = 8 Hz), 7.30 (d, 1 H, H-3, *J*_{H₂-H₃} = 8 Hz); IR (KBr) 3270, 1680 cm⁻¹. Anal. Calcd for C₁₂H₁₁N₃O₃: C, 58.77; H, 4.52; N, 17.13. Found: C, 58.65; H, 4.47; N, 17.20.

1,4-Dihydroxy-2,5-bis(1,3-dimethyl-6-aminouracil-5-yl)-benzene (4): mp >300 °C; EI mass spectrum, *m/e* (relative intensity) 416 (M⁺, 19.1), 399 (39.9), 382 (9.1), 325 (9.6), 286 (9.0), 263 (2.0), 246 (3.9), 57 (100); 100-MHz ¹H NMR (Me₂SO-*d*₆) δ 3.20 (s, 6 H, NCH₃), 3.30 (s, 6 H, NCH₃), 5.80 (s, 4 H, NH₂), 7.90 (s, 2 H, OH), 6.75 (s, 2 H, CH). Anal. Calcd for C₁₈H₂₀N₆O₆: C, 51.92; H, 4.84; N, 20.18. Found: C, 52.05; H, 4.89; N, 20.11.

Fischer-Type Cyclization. **1,3-Dimethyl-5-(2-amino-5-methoxyphenyl)-6-aminouracil (6).** A solution of (*p*-methoxyphenyl)hydrazine hydrochloride (1.75 g, 0.01 mol) and of 1,3-dimethyl-6-chlorouracil (1.74 g, 0.01 mol) in absolute ethanol (50 cm³) was refluxed in the presence of 5.8 g of sodium carbonate for 12 h. The hot solution was filtered. The filtrate was collected and the solvent removed under reduced pressure. The obtained residue was taken up with ether, giving a white precipitate of 6; 63%; mp 242 °C; EI mass spectrum, *m/e* (relative intensity) 277 (M⁺, 21.1), 276 (97.0), 260 (30.0), 259 (100), 244 (16.9), 202 (18.1); 100-MHz ¹H NMR (Me₂SO-*d*₆) δ 3.10 (s, 3 H, NCH₃), 3.30 (s, 3 H, NCH₃), 3.60 (s, 3 H, OCH₃), 4.20 (s, 2 H, NH₂ phenyl), 5.75 (s, 2 H, NH₂ uracil), 6.70 (d, 1 H, H-1, *J*_{H₁-H₂} = 1 Hz), 6.60 (dd, 1 H, H-2, *J*_{H₁-H₂} = 1 Hz, *J*_{H₂-H₃} = 8 Hz), 6.65 (d, 1 H, H-3, *J*_{H₂-H₃} = 8 Hz); IR (KBr) 3350, 3220, 1685 cm⁻¹. Anal. Calcd for C₁₃H₁₆H₄O₃: C, 56.51; H, 5.84; N, 20.28. Found: C, 56.62; H, 5.76; N, 20.26.

1,3-Dimethyl-6-methoxy-9*H*-pyrimido[4,5-*b*]indole-2,4-dione (7). A solution of 6 (2.8 g) in formic acid (50 cm³) was heated at reflux for 1 h, yielding after cooling the pyrimido[4,5-*b*]indole 7 as a white precipitate; 71%; mp 300 °C; EI mass spectrum, *m/e* (relative intensity) 260 (M⁺ + 1, 13.1), 259 (M⁺, 72), 202 (100), 187 (69.9); 100-MHz ¹H NMR (Me₂SO-*d*₆) δ 3.25 (s, 3 H, NCH₃), 3.45 (s, 3 H, NCH₃), 3.75 (s, 3 H, OCH₃), 7.35 (d, 1 H, H-1, *J*_{H₁-H₂} = 1 Hz), 6.75 (dd, 1 H, H-2, *J*_{H₁-H₂} = 1 Hz, *J*_{H₂-H₃} = 8 Hz), 7.30 (d, 1 H, H-3, *J*_{H₂-H₃} = 8 Hz); IR (KBr) 3420, 1690 cm⁻¹. Anal. Calcd for C₁₃H₁₃N₃O₃: C, 60.22; H, 5.05; N, 16.21. Found: C, 59.98; H, 5.01; N, 16.27.

1,3-Dimethyl-6-hydroxy-9*H*-pyrimido[4,5-*b*]indole-2,4-dione (3). A mixture of 7 (1 g) and pyridinium chloride (4 g) was heated at 210 °C for 20 min. The residue was taken up with 50 cm³ of 1 N HCl to give a precipitate of 3 (65%). The analytical data of compounds 3 obtained in this way were identical with those of the sample prepared by the above-described procedure.

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Registry No. 1, 6642-31-5; 2, 78790-67-7; 3, 78790-68-8; 4, 78790-69-9; 5, 6972-27-6; 6, 78790-70-2; 7, 78790-71-3; *p*-benzoquinone, 106-51-4; (*p*-methoxyphenyl)hydrazine hydrochloride, 19501-58-7.

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